Sesterterpenoids from the Marine Sponge Hyrtios erectus

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Four new sesterterpenes, namely, (+)-20-formylhyrtiosal (1), (+)-16-*O*-acetyl-20-formylhyrtiosal (2), 12- α -*O*-acetylhyrtiolide (3), and 5,10-dihydroxyfurospinulosine-1 (4), together with seven known sesterterpenes (5–11) were isolated from the marine sponge *Hyrtios erectus* collected at Hainan, China. The structures were elucidated on the basis of extensive spectroscopic analysis involving mainly one- and two-dimensional NMR as well as mass spectroscopy.

Sponges from the genus Hyrtios occur frequently in tropical and subtropical waters and often form dominant life forms on coral reefs. Previous chemical investigations of different Hyrtios spp. and their associated microorganisms have revealed the presence of numerous structurally unique natural products including scalarane sesterterpenoids,¹⁻¹⁴ acyclic triterpenoids,¹⁵ indole alkaloids,¹⁶⁻¹⁹ macrolides,²⁰⁻²³ and steroids.²⁴ Many of these compounds possess important biological activities. The secondary metabolites of the sponge *Hyrtios erectus* (also called *H. erecta*), collected from Okinawa (Japan) and from Indonesia, have been investigated extensively.^{1–30} However, this sponge when collected from Chinese waters has not been investigated before. In our ongoing research program on bioactive secondary metabolites from marine organisms, H. erectus was collected using scuba on the inner reef (15 m depth) at Hainan Island, South China Sea. Repeated chromatographic separation using silica gel column chromatography followed by reversed-phase HPLC of the CH₂Cl₂ fraction obtained from a MeOH extract of the sponge gave 11 pure compounds (1-11), of which 1-4 were determined as new sesterterpenoids. The structure elucidation of the new compounds mainly employing one- and two-dimensional NMR spectroscopy as well as mass spectroscopy is discussed.

From the 11 natural products isolated from *H. erectus*, compounds **5** and **6** were identical to the known tetracyclic sesterterpenoids furoscalarol and its corresponding deacetyl derivative 12-O-desacetylfuroscalarol, previously reported from the sponge *Cacospongia mollior*,³¹ and the latter two compounds were also isolated from the sponge *Hyrtios* sp.¹⁴ The structures of compounds 7 and 8 were in agreement with the scalarane sesterterpenoids 16-deacetyl-12-episcalarafuran acetate and isoscalarafuran A, respectively, of which the former had been originally isolated from Spongia officinalis,³² whereas the latter had been obtained from S. hispida.33 Compounds 9-11 were identical to scalarane type sesterterpenoids, namely, 12-O-deacetyl-19deoxyscalarin (9),³⁴ scalarin (10),^{31,34} and 12-epideoxoscalarin (11),^{35,36} respectively. All of the known compounds were readily identified by comparison of their spectral data with those reported in the literature.

Compound 1 was obtained as a white amorphous powder, and its molecular formula was determined as $C_{25}H_{36}O_4$ by

HRFABMS. The ¹H and ¹³C NMR spectral features of 1 resembled those of hyrtiosal, a unique rearranged tricyclic sesterterpenoid.³ In the ¹H NMR spectrum of **1**, three downfield protons at δ 7.40 (dd, J = 1.6, 1.5 Hz, H-19), 7.39 (dd, J = 1.5, 0.8 Hz, H-25), and 6.39 (dd, J = 0.8, 1.6 Hz, H-18) were characteristic of a monosubstituted furan moiety, and a small w-bond coupling between H-19 and H-25 through an oxygen bridge was observed. Four methyl singlets at δ 1.23 (3H, s, H-24), 1.09 (3H, s, H-21), 0.95 (3H, s, H-22), and 0.89 (3H, s, H-23) as well as two formyl protons at δ 9.51 (s) and 9.22 (s) indicated that in 1 one of the five methyl groups present in hyrtiosal was replaced by a formyl group instead. The DEPT spectrum of 1 exhibited two methine carbons at δ 206.28 (d) and 205.37 (d), thus supporting the presence of two formyl groups in the molecule. HMBC correlations between δ 9.51 (s, H-12)

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and 19.20 (q, C-24) and between δ 9.22 (s, H-20) and 13.51 (q, C-21) indicated that the second formyl group was positioned at C-4.

The stereochemistry of 1 was assumed to be in agreement with those of (+)-hyrtiosal due to the similar optical rotations between **1** ($[\alpha]_D^{20}$ +60°) and (+)-hyrtiosal ($[\alpha]_D^{20}$ $+64.2^{\circ}$),³⁷ whereas the optical rotation of (-)-hyrtiosal (12) is $[\alpha]^{20}{}_D$ –72.1°.3 Moreover, the NOE correlations of aldehydic proton H-12 (δ 9.51, s) to H-14 (δ 2.07, dd, J = 5.2, 9.4 Hz) and H-24 (δ 1.23, 3H, s), and of H-24 to H-23 (δ 0.89, 3H, s) and H-16 (δ 4.45, dd, J = 5.2, 8.1 Hz), as well as between pairs H-21 (\$\delta\$ 1.09, 3H, s)/H-22 (\$\delta\$ 0.95, 3H, s), H-22/H-23 in the NOESY spectrum, supported the designated configurations of chiral centers at positions C-4, C-10, C-8, C-13, C-14, and C-16. Additionally, compounds related to the (-)-hyrtiosal skeleton featuring a carbonyl group attached to C-4 in β -configuration had been shown to shift the signal of the methyl group at C-4 to more than 24 ppm;³⁸ the chemical shift of C-21 (δ 13.51, q) in **1** also implied a α -configuration for Me-21. Therefore, the structure of the new natural product **1** was determined as a $4-\beta$ formyl-substituted hyrtiosal, named (+)-20-formylhyrtiosal.

Compound **2** was isolated as a minor component, and its molecular formula C27H38O5 was afforded by HR-FABMS, 42 amu higher than that of 1. IR absorptions suggested the presence of two carbonyl groups (1726, 1721 cm⁻¹) and the absence of a hydroxyl group. The ¹H and ¹³C NMR spectra of compound **2** closely resembled those of 1, with the exception of an additional acetyl group that appeared at $\delta_{\rm H}$ 2.07 (s, 3H) and $\delta_{\rm C}$ 21.54 (q), 171.24 (s). The acetoxyl group was assigned to a position at C-16 due to the observed downfield shift of H-16 to δ 5.64 (dd, J =6.0, 7.5 Hz), whereas H-16 of 1 resonated at δ 4.45 (dd. J= 5.2, 8.1 Hz) in comparison. The relative stereochemistry of compound 2 was assumed to be identical to that of 1 due to the similar optical rotation between **1** and **2** ($[\alpha]_D$ +65°), as well as the NOE correlations between H-12 (δ 9.40, s) and the acetyl proton at $\delta_{\rm H}$ 2.07 (3H, s) and between pairs H-24 (\$\delta\$ 1.18, 3H, s)/H-16, H-21 (\$\delta\$ 1.08, 3H, s)/H-22 (\$\delta 0.94, 3H, s), H-22/H-23 (\$\delta 0.92, 3H, s), and H-23/H24 in the NOESY spectrum. Accordingly, compound 2 was identified as (+)-16-O-acetyl-20-formylhyrtiosal.

Compound 3 was isolated as colorless needles. The HRFABMS gave the molecular formula C₂₇H₄₀O₆. IR absorptions suggested the presence of hydroxyl (3421 cm⁻¹), unsaturated δ -lactone (1753, 1632 cm⁻¹), and ester (1716 cm⁻¹) groups. The ¹H NMR spectrum exhibited six methyl groups [& 0.80 (3H, s), 0.81 (6H, s), 0.94 (3H, s), 1.55 (3H, s), and 1.97 (3H, s)] and three protons in the vicinity of the oxygen-bearing substituents [δ 6.60 (s), 5.69 (br s), and 4.94 (d, J = 3.0 Hz)]. The ¹³C NMR and DEPT spectra exhibited 27 carbons, six of which were methyls, seven methylenes, six methines, and eight quaternary carbons (four sp³ and four sp²). The ¹H and ¹³C spectral data were compatible to a large degree with those of the known scalarane sesterterpenoid hyrtiolide, ⁵ with the exception of an additional acetyl group [$\delta_{\rm H}$ 1.97 (3H, s); $\delta_{\rm C}$ 21.02 (q), 170.45 (s)] present in compound 3. The acetyl substituent was assigned to C-12 on the basis of the correlation of H-12 ($\delta_{\rm H}$ 5.69, br) with the acetyl carbon at $\delta_{\rm C}$ 171.12 (s) in the HMBC spectrum. The relative configurations of H-12, H-16, and H-19 were determined by their coupling constants and by interpreting the NOESY spectrum. The NOE correlations of Me-22 (\$\delta\$ 0.80, s)/Me-23 (\$\delta\$ 0.81,s), Me-23/Me-24 (\$\delta\$ 0.94, s), and Me-24/Me-25 (δ 1.55, s) in the NOESY in association with ¹³C NMR chemical shifts of the methyl groups ⁵ indicated *trans* junctures of the rings A–D, identical to those of hyrtiolide. The broad signals of H-12 (δ 5.69, br s) and H-16 (δ 4.94, br d, J = 3.0 Hz) suggested that H-12 and H-16 are in equatorial positions. The NOE correlations between H-12/Me-25 ($\delta_{\rm H}$ 1.55, s), Me-25/H-15-($\delta_{\rm H}$ 2.12(d), and H-15/H-16 in the NOESY spectrum further confirmed the β -configurations for H-12 and H-16. The α -configuration of the hydroxyl group at C-19 was deduced by NOE correlations between H-19 (δ 6.60, br) and H-12 and between H-19 and Me-25. Thus, compound **3** was identified as 12- α -O-acetylhyrtiolide.

Compound 4 was isolated as a white oil, and its molecular formula, C₂₅H₃₈O₃, was established on the basis of the HRFABMS spectrum. The ¹³C NMR and DEPT spectra exhibited five methyl groups [δ 26.55 (q, C-25), 17.70 (q, C-24), 16.02 (q, C-19), 16.33 (q, C-14), 12.20 (q, C-9)], six methylene carbons [8 25.71 (t, C-21), 39.73 (t, C-20), 26.76 (t, C-16), 39.85 (t, C-15), 34.13 (t, C-11), 36.30 (t, C-6)], nine methines [8 124.35 (d, C-22), 123.96 (d, C-17), 119.69 (d, C-12), 120.95 (d, C-7), 139.04 (d, C-4), 108.58 (d, C-2), 143.29 (d, C-1), 77.02 (d, C-10), 66.66 (d, C-5)], and five quaternary carbons [8 140.53 (s, C-8), 138.90 (s, C-13), 131.34 (s, C-23), 135.31 (s, C-18), 128.62 (s, C-3)], of which four (C-1 to C-4) were attributed to a β -substituted furan moiety. In ¹H NMR spectrum, three proton signals at δ 7.41 (d, J = 1.0 Hz, H-1), 7.41 (br s, H-4), and 6.44 (br s, H-2) were assigned for the β -substituted furan moiety. The proton signals at δ 5.47 (dd, J = 7.0, 7.0 Hz, 1H, H-7) and 5.11-5.08 (m, 3H, H-12, H-17, H-22) were due to four olefinic protons. Additionally, there were five methyl singlets at δ 1.70 (3H, s, H-25), 1.68 (3H, s, H-9), 1.66 (3H, s, H-14), and 1.62 (6H, s, H-19, H-24). ¹³C NMR data indicated the presence of four trisubstituted olefinic bonds in addition to the β -substituted furan ring [δ 108.58 (d, C-2), 128.62 (s, C-3), 139.04 (d, C-4), 143.29 (d, C-1)]. Since the seven sites of unsaturation that had been revealed by interpretation of the mass spectrum could be accounted for by the four double bonds and the furan ring, compound 4 was identified as a linear sesterterpene. The HMQC and HMBC spectral analysis enabled us to establish the gross structure, which is closely related to a known linear sesterterpene, furospinulosin-1.36 The new natural product 4 differed from the known furospinulosin-1 by the presence of two additional hydroxyl groups, which were assigned to C-5 and C-10, respectively, on the basis of HMBC correlations observed between H-4 (δ 7.41, br) and C-5 (δ 66.66, d) and between Me-9 and C-10, and Me-9 was further correlated with the olefinic carbons C-8 and C-7. The geometry of the double bonds was determined to be all Eon the basis of the chemical shifts of the methyl groups at high field (<18 ppm) and a downfield methyl signal at δ 26.50 (q, C-25) being cis to the olefinic proton H-22.36,39 Accordingly, the structure of 4 was determined to be 5,-10-dihydroxylfurospinulosine-1.

Experimental Section

General Experimental Procedures. Melting points were measured by a XT-4A micromelting point apparatus without correction. The IR spectra were determined on a Thermo Nicolet Nexus 470 FT-IR spectrometer. ¹H NMR and ¹³C NMR as well as 2D NMR spectra were recorded on an AVANCE-500 FT 500 MHz NMR spectrometer using TMS as internal standard. EIMS spectra were performed with a Bruker APEX II mass spectrometer. Column chromatography was performed with silica gel (200–300 mesh). HF254 silica gel for TLC was provided by Sigma Co. Ltd. Sephadex LH-20 (18–110 μ m) was obtained from Pharmacia Co.

Sponge Material. The specimen of *Hyrtios erectus* was collected from Hainan Island, South China Sea, in July 2002. The species was identified by Dr. R. van Soest (Institute of

Table 1. ¹³C and ¹H NMR Data of Compounds 1, 2, and Hyrtiosal (12)^a

	1		2		12 ³	
no.	δ_{C}	$\delta_{\rm H}$ (J Hz)	δ_{C}	$\delta_{\rm H}$ (J Hz)	δ_{C}	$\delta_{ m H}$ (JHz)
1	39.20 CH ₂	1.01 m	39.53 CH ₂	1.03 m	40.2 CH ₂	
	~	1.53 m	~	1.53 m		
2	16.54 CH ₂	1.61 m	17.10 CH ₂	1.55 m	18.8 CH ₂	
		1.72 m		1.69 m	~	
3	32.49 CH ₂	1.25, 1.28 m	30.02 CH ₂	1.28 m	42.4 CH_2	
3 4 5	49.30 qC		50.44 qC [~]		33.1 qC ~	
5	49.58 ĈH	1.43 m	49.71 ĈH	1.41 m	57.4 ĈH	0.85 m
6	21.49 CH ₂	0.98 m	21.62 CH ₂	1.02 m	18.3 CH ₂	1.40 m
		1.42 m		1.49 m		1.60 m
7	39.62 CH ₂	1.19 m	40.44 CH ₂	1.13 m	40.3 CH ₂	1.13 m
		1.72 m		1.78 m		1.72 td (3.2, 12.5)
8	44.79 gC		42.56 gC		44.5 qC	
8 9	60.05 CH	1.18 dd (<i>7.0</i>)	60.13 CH	1.16 d (<i>7.9</i>)	60.4 CH	1.07 dd (6.2, 13.8)
10	35.72 qC		36.21 qC		36.8 qC	(, , , , , , , , , , , , , , , , , , ,
11	33.64 CH ₂	1.69 m	33.55 CH ₂	1.35 m	33.68 CH ₂	1.40 m
		1.97 dd (6.0, 11.2)		1.91 dd (<i>6.2</i> , <i>13.0</i>)	00100 01-2	1.89 dd (<i>6.3, 13.1</i>)
12	206.28 CH	9.51 s	204.98 CH	9.40 s	205.7 CH	9.45 s
13	52.69 qC		53.36 qC		52.8 qC	
14	48.13 CH	2.07 dd (5.2, 9.4)	49.12 CH	1.78 m	48.1 CH	1.98 dd (6.6, 7.9)
15	33.49 CH ₂	1.47 m	33.86 CH ₂	1.43 m	33.69 CH ₂	1.62 m
		1.56 m	<i>L</i>	1.55 m	<u>L</u>	
16	64.25 CH	4.45 dd (5.2, 8.1)	66.97 CH	5.64 dd (<i>6.0, 7.5</i>)	64.2 CH	4.42 dd (6.0, 7.4)
17	129.19 qC		124.5 gC		129.2 qC	
18	108.44 CH	6.39 dd (0.8, 1.6)	109.13 CH	6.37 br s	109.5 CH	6.37 t (<i>1.1</i>)
19	143.29 CH	7.40 dd (1.6, 1.5)	143.65 CH	7.40 d (<i>1.6</i>)	143.2 CH	7.36 br s
20	205.37 CH	9.22 s	208.22 CH	9.23 s	33.5 CH	0.845 s
21	13.51 CH ₃	1.09 s	13.89 CH ₃	1.08 s	21.2 CH ₃	0.82 s
22	15.74 CH ₃	0.95 s	16.14 CH ₃	0.94 s	15.7 CH ₃	0.86 s
23	16.54 CH ₃	0.89 s	16.50 CH ₃	0.92 s	16.5 CH ₃	0.851 s
24	19.20 CH ₃	1.23 s	19.61 CH ₃	1.18 s	19.1 CH ₃	1.18 s
25	138.80 CH	7.39 dd (<i>1.5, 0.8</i>)	143.71 CH	7.42 br s	139.8 CH	7.36 br s
Ãc		(110) 010)	21.54 CH ₃	2.07 s		
			171.24 gC			

^a The ¹H and ¹³C NMR spectra are measured in CDCl₃.

Systematic Population Biology, Amsterdam University, The Netherlands). A voucher specimen (HS-17) is deposited at the State Key Laboratory of Natural and Biomimetic Drugs, Peking University.

Extraction and Isolation. The sponge (2.3 kg, wet wt) was homogenized and then extracted with MeOH and CHCl₃ successively. The concentrated total extract was partitioned between petroleum ether and 90% MeOH to remove lipids, and then the 90% aqueous MeOH was concentrated under reduced pressure to afford a degreased extract, which was partitioned between H₂O and CH₂Cl₂. CH₂Cl₂ extract was subjected to silica gel column chromatography eluted with an equivalent petroleum ether-ethyl acetate stepwise gradient to obtain 10 fractions. Fraction 3 (petroleum ether-EtOAc, 3:1) was subjected to a Sephadex LH-20 column and eluted with 85% aqueous MeOH to get fractions 2-1 to 2-7. Fraction 2-1 was chromatographed on silica gel eluting with petroleum ether-EtOAc (2:1) to yield compounds 1 (7.8 mg), 2 (2.3 mg), 5 (1263 mg), 7 (4.3 mg), 8 (158 mg), and 11 (28.0 mg). Fractions 2-3 to 2-5 were combined and subjected to a Sephadex LH-20 column and eluted with 80% MeOH to collect 10 fractions, which were monitored by TLC and ¹H NMR spectra. Terpenoid-containing fractions were combined and separated by silica gel column chromatography using petroleum ether-acetone (2:1) as eluting solvent to give compounds 3 (5.2 mg), 4 (11.4 mg), 6 (87 mg), 9 (2.1 mg), and 10 (821 mg).

Compound (1): white amorphous; $[\alpha]^{25}_{D}$ +60° (*c* 0.245, MeOH); IR (Nujol) ν_{max} 3432 (br), 2927, 2870, 1718, 1502, 1455, 1386, 1159, 1054, 1023, 874, 796, 730, 601 cm⁻¹; ¹H and ¹³C NMR (CDCl₃, 500 MHz), see Table 1; ESIMS *m/z* (positive mode) 401 [M + H]⁺, 383 [M - OH]⁺, 367, 337, 273; HR-FABMS *m/z* 423.24972 (calcd for C₂₅H₃₆O₄Na, 423.25113).

Compound (2): white oil, $[\alpha]^{25}_{D}$ +65° (*c* 0.2, MeOH); IR (Nujol) ν_{max} 2929, 2869, 1725, 1721, 1504, 1459, 1372, 1239, 1161, 1023, 971, 874, 796, 729, 602 cm⁻¹; ¹H NMR (CDCl₃, 500 MHz), see Table 1; ¹³C NMR (CDCl₃, 125 MHz), see Table 1; FABMS *m*/*z* (positive mode) 443 [M + 1]⁺, 383 [M - OAc]⁺, 399 [M - Ac]⁺; HRFABMS *m*/*z* 465.26058 (calcd for C₂₇H₃₈O₅-Na, 465.26169).

Table 2. ¹³C and ¹H NMR Data of 4 and 3

		4 ^a	3^b		
no.	$\delta_{\rm C}$	$\delta_{ m H}$ (J Hz)	$\delta_{\rm C}$	$\delta_{\mathrm{H}} \left(J \mathrm{Hz} \right)$	
1	143.29 CH	7.41 d (<i>1.0</i>)	40.07 CH ₂	1.57, 0.75 m	
2 3	108.58 CH	6.44 br s	18.32 CH ₂	1.47, m	
3	128.62 qC		42.27 CH ₂	1.34,1.08 m	
4	139.04 ĈH	7.41 br s	33.38 qC		
5	66.66 CH	4.73 dd (6.8, 7.1)	56.60 ĈH	0.82 m	
6	36.30 CH ₂	2.51, 2.55 m	18.83 CH ₂	1.42, 1.33 m	
7	120.95 CH	5.47 dd (<i>7.0, 7.0</i>)	41.50 CH ₂	1.84, 1.20 m	
8	140.53 qC		37.73 qC		
9	12.20 ČH ₃	1.68 s	53.21 ČH	1.54 m	
10	77.02 CH	4.04 dd (<i>6.1, 6.8</i>)	37.22 qC		
11	34.13 CH ₂	2.26, 2.30 m	21.99 ČH ₂	2.09, 1.84 m	
12	119.69 CH	5.10 m	75.49 CH	5.69 br s	
13	138.90 qC		42.12 qC		
14	16.33 ČH ₃	1.66 s	45.93 ČH	2.51 dd (<i>12.5</i>)	
15	39.85 CH ₂	2.03 m	28.18 CH ₂	2.13 d (14.0) 1.90	
				ddd (14.0, 4.0)	
16	26.76 CH ₂	2.06 m	59.53 CH	4.94 d (<i>3.0</i>)	
17	123.96 CH	5.10 m	131.42 qC		
18	135.31 qC		168.10 qC		
19	16.02 ĈH ₃	1.62 s	98.27 ĈH	6.60 s	
20	39.73 CH ₂	2.11 m	171.12 qC		
21	25.71 CH ₂	1.99 m	33.25 ČH ₃	0.81 s	
22	124.35 CH	5.10 m	21.48 CH3	0.80 s	
23	131.34 qC		17.57 CH ₃	0.81 s	
24	17.70 CH_3	1.62 s	16.12 CH ₃	0.94 s	
25	26.55 CH ₃	1.70 s	19.22 CH ₃	1.55 s	
Ac			21.02 CH ₃	1.97 s	
			170.45 qC		

^{*a*} The ¹H and ¹³C NMR spectra are measured in pyridine- d_5 . ^{*b*} The ¹H and ¹³C NMR spectra are measured in CDCl₃.

Compound (3): colorless needles, mp 220–222 °C, $[\alpha]^{25}_{\rm D}$ +81.46° (*c* 0.205, MeOH); IR (Nujol) 3421(br), 2931, 1753, 1716, 1632, 1463, 1376, 1238 cm⁻¹; ¹H and ¹³C data, see Table 2, ESIMS *m/z*. 460 [M]⁺, 920 2M⁺; HRFABMS *m/z* 483.27315 (calcd for C₂₇H₄₀O₆Na, 483.27226).

Compound (4): colorless oil, [α]²⁵_D +32.52° (*c* 0.12, MeOH); IR (Nujol) 3383 (br), 2925, 1453, 1378, 1026, 974, 875, 842 cm⁻¹; ¹H and ¹³C NMR data, see Table 2; ESIMS m/z 404 [M $+ NH_4$]⁺, 369 [M - OH]⁺, 351 [M - 2OH]⁺; HRFABMS m/z 385.27441 (calcd for C25H37O3, 385.27427).

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Supporting Information Available: This material is available free of charge via the Internet at http://pubs.acs.org.

References and Notes

- Kobayashi, M.; Okamoto, T.; Hayashi, K.; Yokoyama, N.; Sasaki, T.; Kitagawa, I. *Chem. Pharm. Bull.* **1994**, *42*, 265–270.
 Bourguet-Kondracki, M. L.; Martin, M. T.; Debitus, C.; Guyot, M. *Tetrahedron Lett.* **1994**, *35*, 109–110.
- (3)Iguchi, K.; Shimada, Y.; Yamada, Y. J. Org. Chem. 1992, 57, 522-524.
- (4) Crews, P.; Bescansa, P.; Bakus, G. J. Experientia 1985, 41, 690-691.
- (5)Miyaoka, H.; Nishijima, S.; Mitome, H.; Yamada, Y. J. Nat. Prod. 2000, 63, 1369-1372.
- (6) Ryu, G.; Matsunaga, S.; Fusetani, N. J. Nat. Prod. 1996, 59, 515-5Ĭ7.
- (7) Crews, P.; Bescansa, P. J. Nat. Prod. 1986, 49, 1041-1052.
- Tsuchiya, N.; Sato, A.; Hata, T.; Sato, N.; Sasagawa, K.; Kobayashi, T. *J. Nat. Prod.* **1998**, *61*, 468–473. Kirsch, G.; König, G. M.; Wright, A. D.; Kaminsky, R. *J. Nat. Prod.* (8) (9)
- 2000, 63, 825-829. (10) Fontana, A.; Cavaliere, P.; Ungur, N.; D'Souza, L.; Parameswaram,
- P. S.; Cimino, G. J. Nat. Prod. 1999, 62, 1367-1370. (11) Pettit, G. R.; Cichacz, Z. A.; Tan, R.; Herald, D. L.; Melody, N.; Hoard,
- M. S.; Doubek, D. L.; Hooper, J. N. A. Collect. Czech. Chem. Commun. **1998**, 63, 1671-1677.
- (12)Pettit, G. R.; Tan, R.; Melody, N.; Cichacz, Z. A.; Herald, D. L.; Hoard, M. S.; Pettit, R. K.; Chapuis, J. C. Bioorg. Chem. Lett. 1998, 8, 2093-2098.
- (13) Pettit, G. R.; Cichacz, Z. A.; Tan, R.; Hoard, M. S.; Melody, N.; Pettit, R. K. *J. Nat. Prod.* **1998**, *61*, 13–16.
 (14) Doi, Y.; Shigemori, H.; Ishibashi, M.; Mizobe, F.; Kawashiba, A.;
- Nakaike, S.; Kobayashi, J. Chem. Pharm. Bull. 1993, 41, 2190–2191.

- (15) Williams, D. E.; Tahir, A.; Andersen, R. J. J. Nat. Prod. 1999, 62, 653 - 654
- (16) Kobayashi, J.; Murayama, T.; Ishibashi, M.; Kosuge, S.; Takamatsu, M.; Ohizumi, Y.; Kobayashi, H.; Ohta, T.; Nozoe, S.; Sasaki, T. Tetrahedron 1990, 46, 7699-7702.
- (17) Kobayashi, M.; Aoki, S.; Gato, K.; Matsunami, K.; Kurosu, M.; Kitagawa, J. Chem. Pharm. Bull. 1994, 42, 2449–2451. (18) Bourgouet-Kondracki, M. L.; Martin, M. T.; Guyot, M. Tetrahedron
- Lett. 1996, 37, 3457-3460.
- (19) Varoglu, M.; Corbett, T. H.; Valeriote, F. A.; Crews, P. J. Org. Chem. 1997, 62, 7078-7079.
- (20) Kobayashi, M.; Aoki, S.; Sakai, H.; Kawazoe, K.; Kihara, N.; Sasaki, T.; Kitagawa, I. Tetrahedron Lett. 1993, 34, 2795-2798.
- (21) Kobayashi, M.; Aoki, S.; Sakai, H.; Kihara, N.; Sasaki, T.; Kitagawa, I. Chem. Pharm. Bull. 1993, 41, 989-991.
- (22) Kobayashi, M.; Aoki, S.; Kitagawa, I. Tetrahedron Lett. 1994, 35, 1243-1246.
- (23) Kobayashi, M.; Aoki, S.; Gato, K.; Kitagawa, I. Chem. Pharm. Bull. 1996, 44, 2142-2149.
- (24) Koch, P.; Djerassi, C.; Lakshmi, V.; Schmitz, F. J. Helv. Chem. Acta **1983**, 66, 2431-2436.
- (25) Endo, M. Baiosaiensu to Indasutori 1989, 47, 262-265.
- (26) Saxton, J. E. Nat. Prod. Rep. 1992, 9, 393-446.
 (27) Endo, M.; Nakagawa, M.; Hamamoto, Y.; Ishihama, M. Pure Appl. Chem. 1986, 58, 387-394.
- (28) Molina, P.; Fresneda, P. M.; Garcia-Zafra, S. Tetrahedron Lett. 1996, 37, 9353-9356.
- (29) Ovechkina, Y. Y.; Pettit, R. K.; Cichacz, Z. A.; Pettit, G. R.; Oakley, B. R. Antimicrob. Agents Chemother. 1999, 43, 1993-1999.
- Basabe, P.; Diego, A.; Diez, D. marcos I. S.; Urones, J. G. Synlett. 2000, 1807–1809. (30)
- Cimino, G.; Cafieri, F.; De Napoli, L.; Fattorusso, E. Tetrahedron Lett. (31)**1978**, *23*, 2041–2044.
- (32) De Giulio, A.; De Rosa, S.; Di Vincenzo, G.; Zavodnik, N. J. Nat. Prod. 1989, 52, 1258-1262
- (33) Davis, R.; Capon, R. J. Aust. J. Chem. 1993, 46, 1295-1299. (34)Fattorusso, E.; Magno, S.; Santacroce, C. Sica, D. Tetrahedron 1972, 28. 5993-5997
- (35) Cimino, G.; De Stefano, S.; Minale, L.; Trivellone, E. J. Chem. Soc., Perkin Trans. 1 1977, 1587-1593.
- Walker, R. P.; Thompson, J. E.; Faulkner, J. J. Org. Chem. 1980, 45, (36)4976 - 4979
- Lunardi, I.; Santiago, G. M. P.; Imamura, P. M. Tetrahedron Lett. (37)2002, 43, 3609-3611.
- (38)Salva, J.; Faulkner, D. J. J. Org. Chem. 1990, 55, 1941-1943.
- (39) Asari, F.; Kusumi, T.; Kakisawa, H. J. Nat. Prod. 1989, 52, 1167-1169.

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